

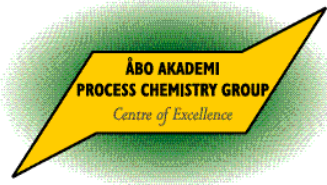
# Ion-selective electrodes & ion-sensors - engineering analytical tools

...electrochemical sensors in routine clinical chemistry

**Andrzej Lewenstam**



***AGH - University of Science  
and Technology  
Krakow, Poland***



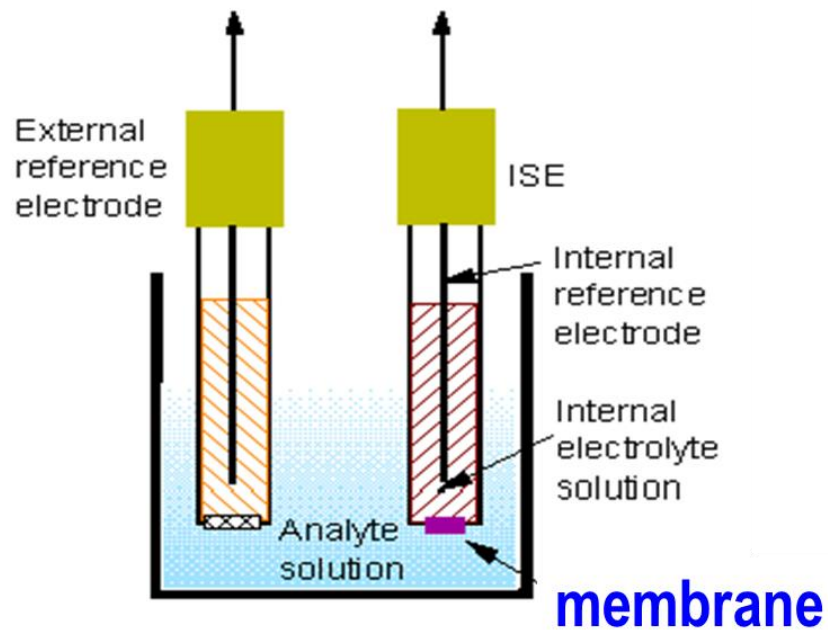
***Åbo Akademi University  
Åbo -Turku, Finland***



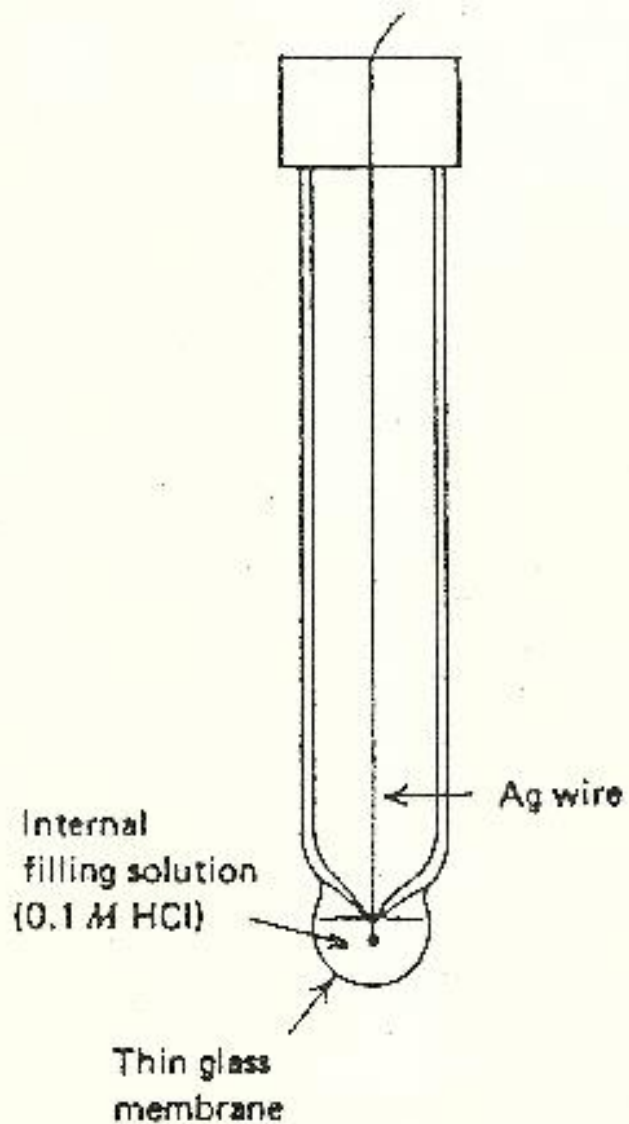
**ThermoFisher**  
SCIENTIFIC

***Helsinki, Finland / Boston, USA***

# GALVANIC CELL

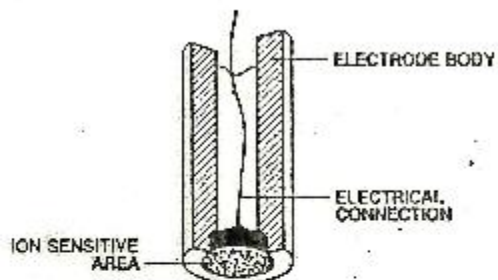


# Ion selective electrodes (ISEs): WITH A GLASS MEMBRANE

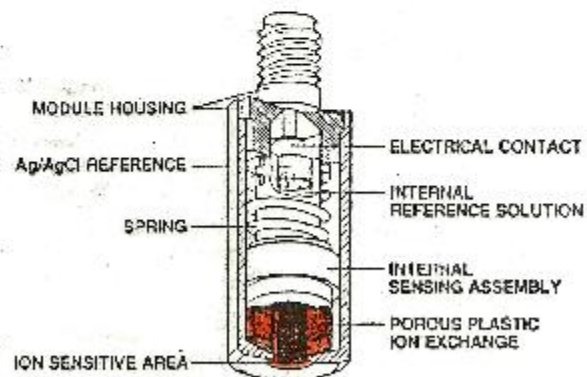


Haber Klemensiewicz 1909

# ISEs with SOLID-STATE and PLASTIC MEMBRANES



Solid State Electrode (Cu<sup>2+</sup>)



Liquid Membrane Electrode (Module) (Cu<sup>2+</sup>)

## Active materials of electrode membranes

Primary ion	Orion electrode	Other homogeneous membranes	Heterogeneous membranes
F <sup>-</sup>	LaF <sub>3</sub>		
Cl <sup>-</sup>	AgCl/Ag <sub>2</sub> S	Hg <sub>2</sub> Cl <sub>2</sub> /HgS, AgCl	AgCl
Br <sup>-</sup>	AgBr/Ag <sub>2</sub> S	Hg <sub>2</sub> Br <sub>2</sub> /HgS, AgBr	AgBr
I <sup>-</sup>			
CN <sup>-</sup>	AgI/Ag <sub>2</sub> S	AgI	AgI
Hg <sub>2</sub> <sup>2+</sup>			
SCN <sup>-</sup>	AgSCN/Ag <sub>2</sub> S	Hg(SCN) <sub>2</sub> /HgS, AgSCN	AgSCN
S <sup>2-</sup>	Ag <sub>2</sub> S	Ag <sub>2</sub> S	Ag <sub>2</sub> S
Cu <sup>2+</sup>	Cu <sub>2</sub> S/Ag <sub>2</sub> S	CuSe	Cu <sub>2</sub> S, Cu <sub>2</sub> S/Ag <sub>2</sub> S
Pb <sup>2+</sup>	PbS/Ag <sub>2</sub> S		PbS/Ag <sub>2</sub> S
Cd <sup>2+</sup>	CdS/Ag <sub>2</sub> S		CdS/Ag <sub>2</sub> S

## Electrodes based on ion exchangers

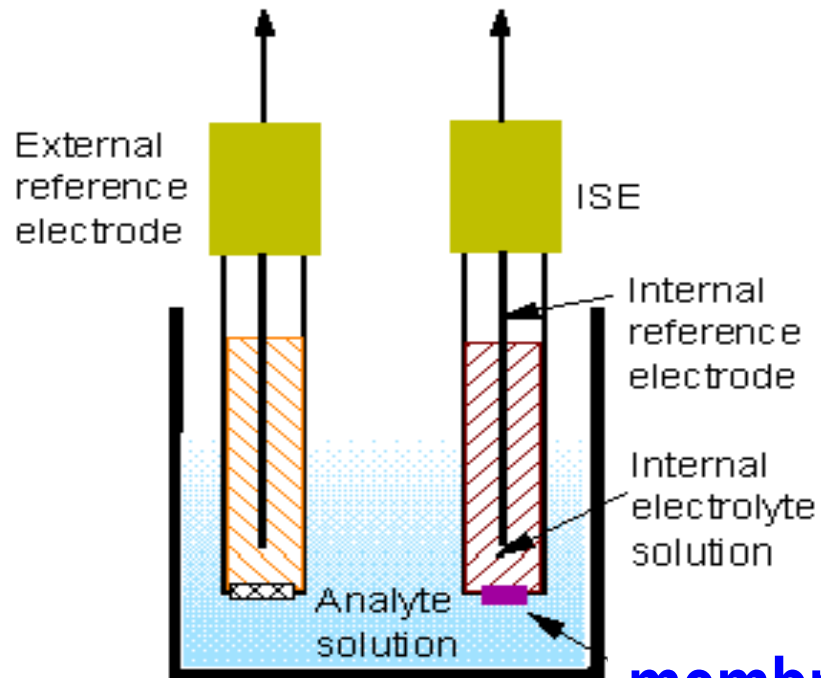
Determinand	Form of membrane	Active material	Solvent or plasticizer
Ca <sup>2+</sup>	(a) liquid	calcium di-(n-decyl) phosphate	di-(n-octylphenyl) phosphonate
	(b) solid (PVC)	calcium di-(n-decyl) phosphate	di-(n-octylphenyl) phosphonate
	(c) solid (PVC) 'Telectrode'	calcium di-(n-octylphenyl) phosphate	di-(n-octylphenyl) phosphonate
NO <sub>3</sub> <sup>-</sup>	(a) liquid	tridodecylhexadecylammonium nitrate	n-octyl-2-nitrophenyl ether
	(b) solid (PVC)	tridodecylhexadecylammonium nitrate	n-octyl-2-nitrophenyl ether
	(c) liquid	tris(substituted 1,10-phenanthroline) nickel(II) nitrate	p-nitrocymene
	(d) solid PVC	tris(substituted 1,10-phenanthroline) nickel(II) nitrate	p-nitrocymene
	(e) liquid	tris(substituted 1,10-phenanthroline) nickel(II) nitrate	p-nitrocymene
ClO <sub>4</sub> <sup>-</sup>	liquid	tris(substituted 1,10-phenanthroline) iron(III) perchlorate	p-nitrocymene
BF <sub>4</sub> <sup>-</sup>	liquid	tris(substituted 1,10-phenanthroline) nickel(II) tetrafluoroborate	p-nitrocymene
Divalent cations ('water hardness')	liquid	calcium di-(n-decyl) phosphate	decanol
CO <sub>3</sub> <sup>2-</sup>	liquid	tri(n-octyl)methylammonium chloride	trifluoro-n-octyl-p-butylbenzene
U (VI)	(a) liquid	methylene blue-uranyl tribenzoate	o-dichlorobenzene
	(b) solid (PVC)	di(2-ethylhexyl) phosphoric acid	diamylamyl phosphonate
Cl <sup>-</sup>	liquid	dimethyl-diocetadecylammonium chloride	?

## Electrodes based on neutral carriers

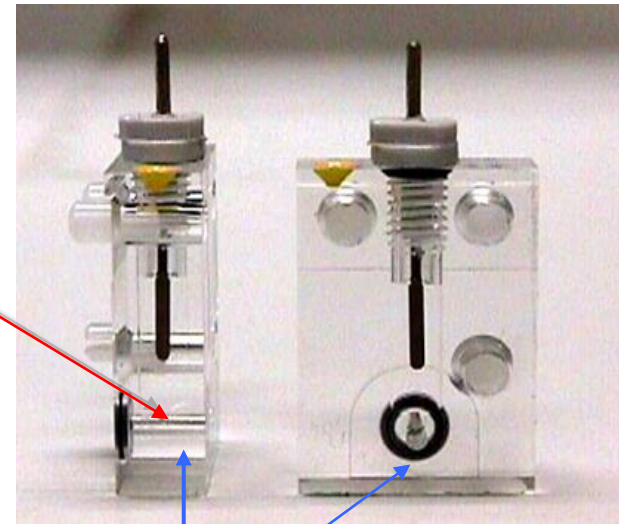
Determinand	Form of membrane	Active material	Solvent or plasticizer
K <sup>+</sup>	(a) liquid	valinomycin	diphenyl ether
	(b) solid (PVC)	valinomycin	dioctyladipate
	(c) solid (silicone rubber)	valinomycin	—
NH <sub>4</sub> <sup>+</sup>	liquid	nonactin/momactin	tris(2-ethylhexyl) phosphate
Ca <sup>2+</sup>	solid (PVC)	see Ref. 18	o-nitrophenyl octyl ether
Ba <sup>2+</sup>	liquid	nonylphenoxy poly(ethyleneoxy) ethanol	p-nitroethylbenzene

Ba<sup>2+</sup> (tetraphenylborate)<sub>2</sub>

# ISE electrodes

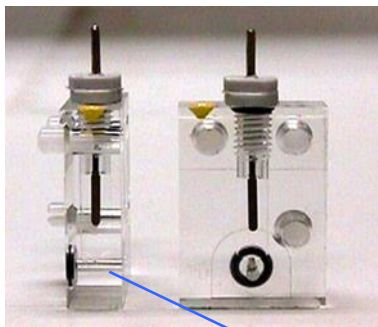


membrane

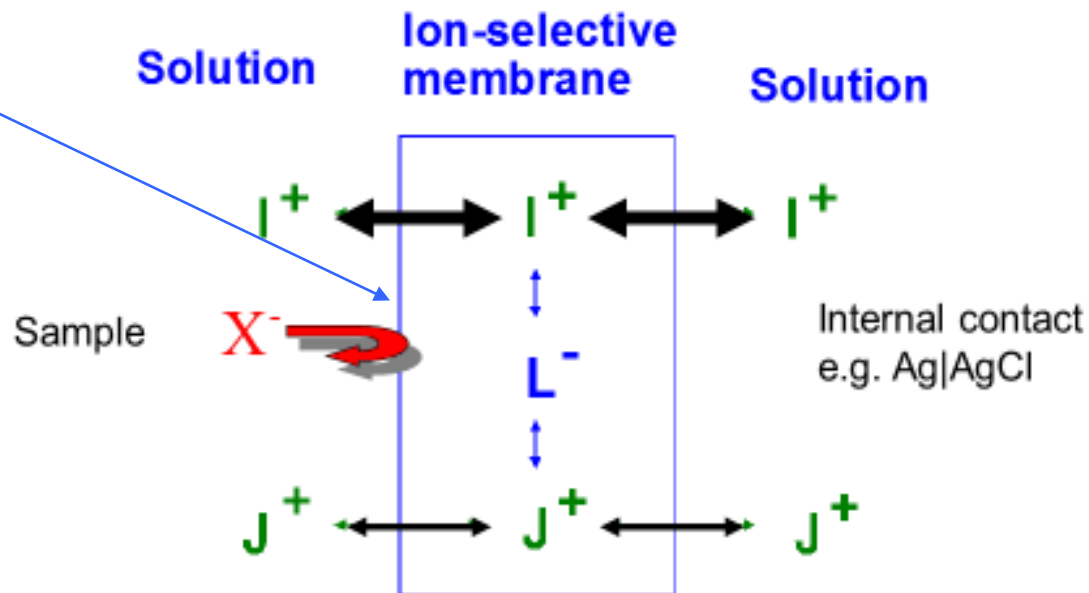


flow-through channel

2.5 μL



# Membrane potential formation process



(1) charge separation at the membrane-solution interface is the prerequisite of a membrane potential

# ...ion-sensor response

(under equilibrium)

$$E = E^0 + \frac{RT}{z_i F} \ln(a_i)$$



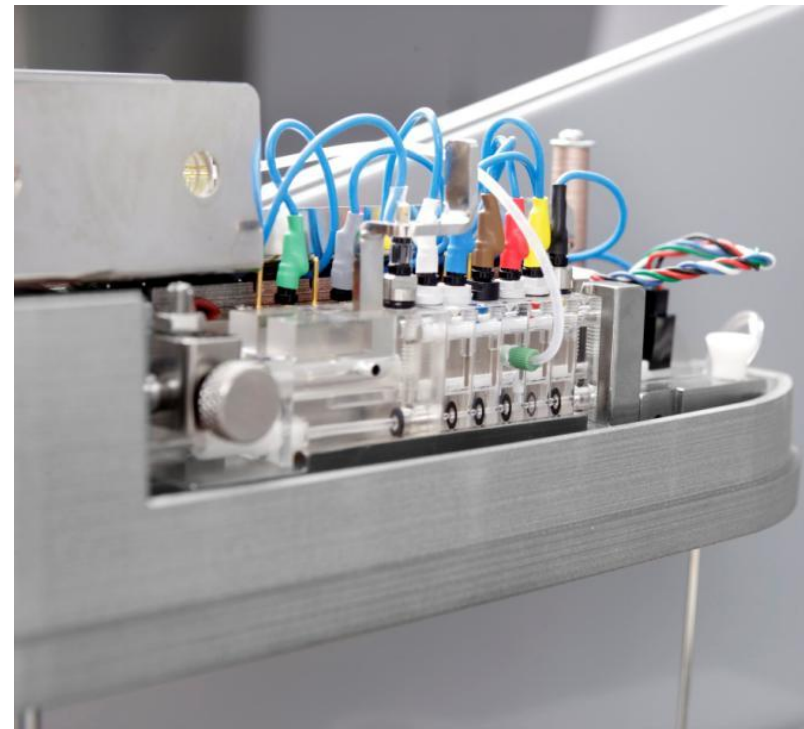
W. H. NERNST  
1864-1941

$$E = \text{const} + \frac{RT}{z_i F} \ln \left[ a_i + \sum_{j \neq i} K_{ij}^{\text{Pot}} (a_j)^{z_i/z_j} + L \right]$$



# Ion-Selective Electrodes (ISEs) unit in clinical analyzer

- Direct measurements  
**Na<sup>+</sup>, K<sup>+</sup>, Cl<sup>-</sup>** (pH, Ca<sup>2+</sup>, Mg<sup>2+</sup>, Li<sup>+</sup>, TCO<sub>2</sub>)
- **Maintenance-free electrodes**,  
delivered as ready-to-use
- **Response time: 2.5 sec**
- **Reproducibility: +/- 0.1 mV**
- **Life-time 10000 samples**
- **Shelf life-time 1 year**





DOI: 10.1002/elan.201400061

## Routines and Challenges in Clinical Application of Electrochemical Ion-Sensors

Andrzej Lewenstam <sup>\*,[a, b]</sup>

**Abstract:** This review provides information on application of potentiometric ion-sensors in routine clinical analysis. The text comprises sensor design, way-of-use, conventions and practical recommendations, and elements of IS re-

sponse theory. An instrument producer point of view, based on direct industrial involvement of the author, is emphasized.

**Keywords:** Clinical analysis • Electrolytes in blood • Ionized magnesium • Ion-sensors • Solid-contact

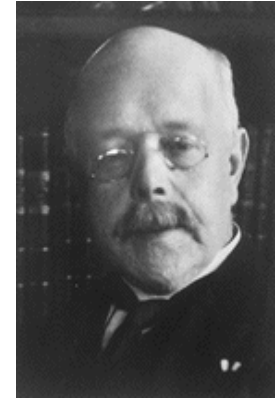


# CHALLENGES

## ...ion-sensor response

(under equilibrium)

$$E = E^0 + \frac{RT}{z_i F} \ln(a_i)$$



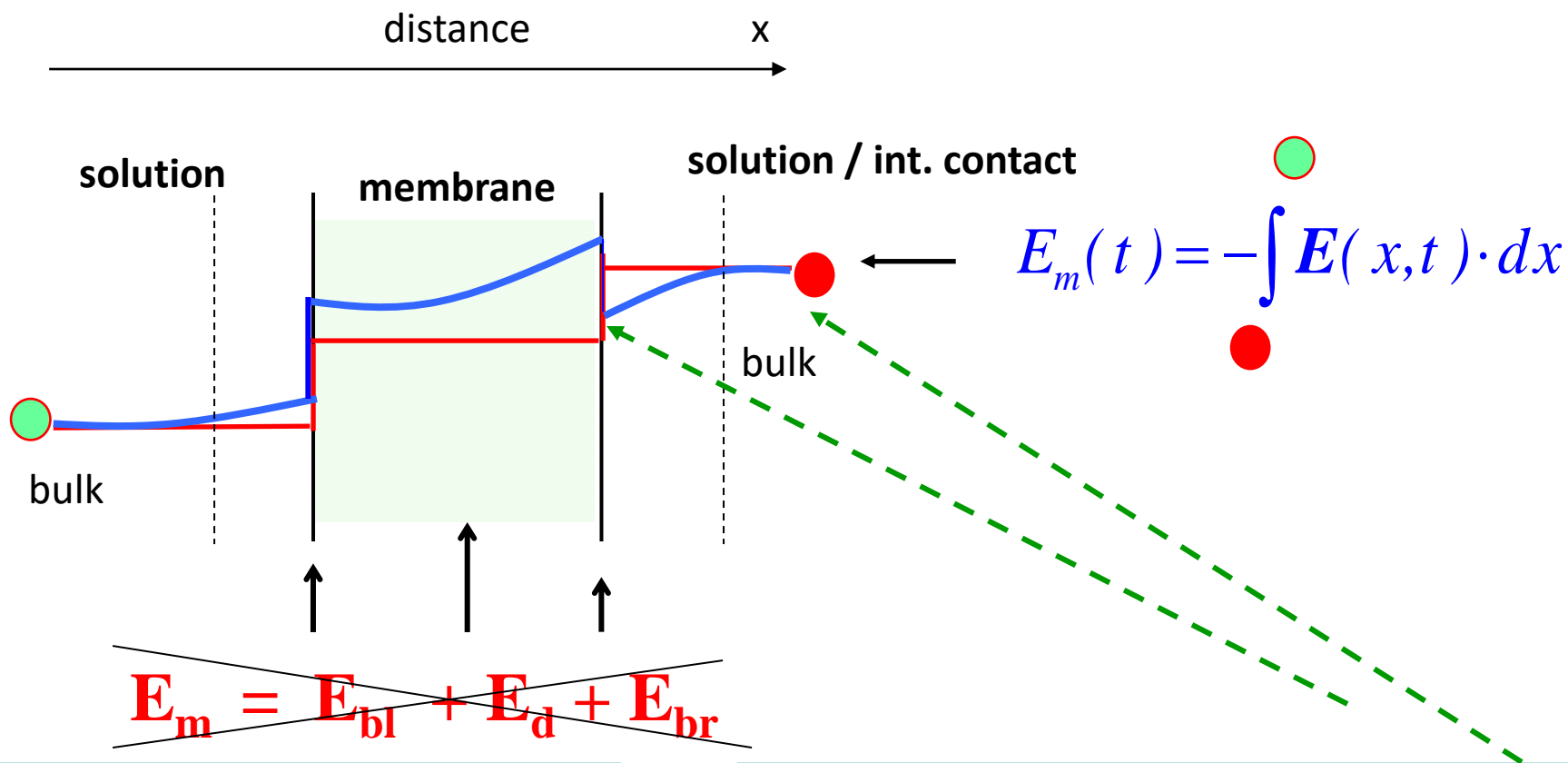
W. H. NERNST  
1864-1941

But....

$$E = \text{const} + \frac{RT}{z_i F} \ln \left[ a_i + \sum_{j \neq i} K_{ij}^{\text{Pot}} (a_j)^{z_i/z_j} + L \right]$$

← time-dependant →

# Equilibrium vs. advanced modelling of membrane potential



Electroneutrality is not assumed

Equilibrium is not assumed ( $c_{\text{surface}} \neq c_{\text{bulk}}$ )

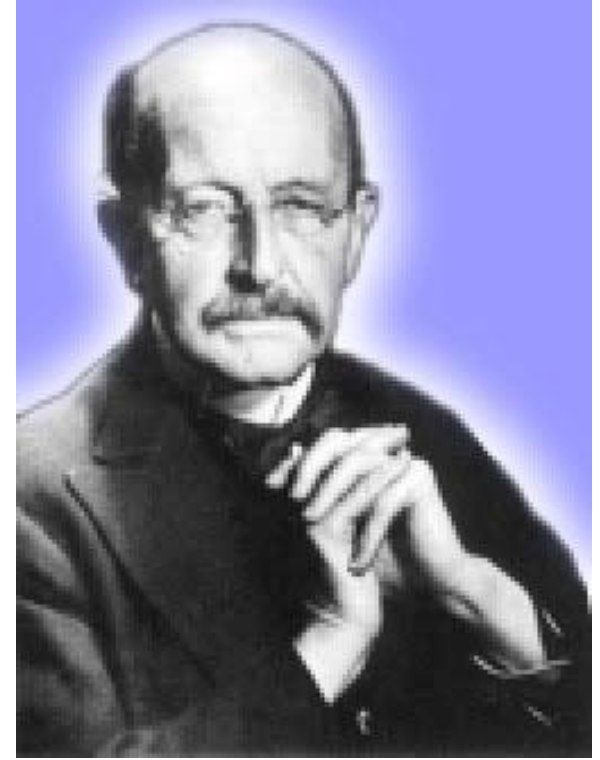
# NPP Model



Siméon Denis **POISSON**  
1781 -1840

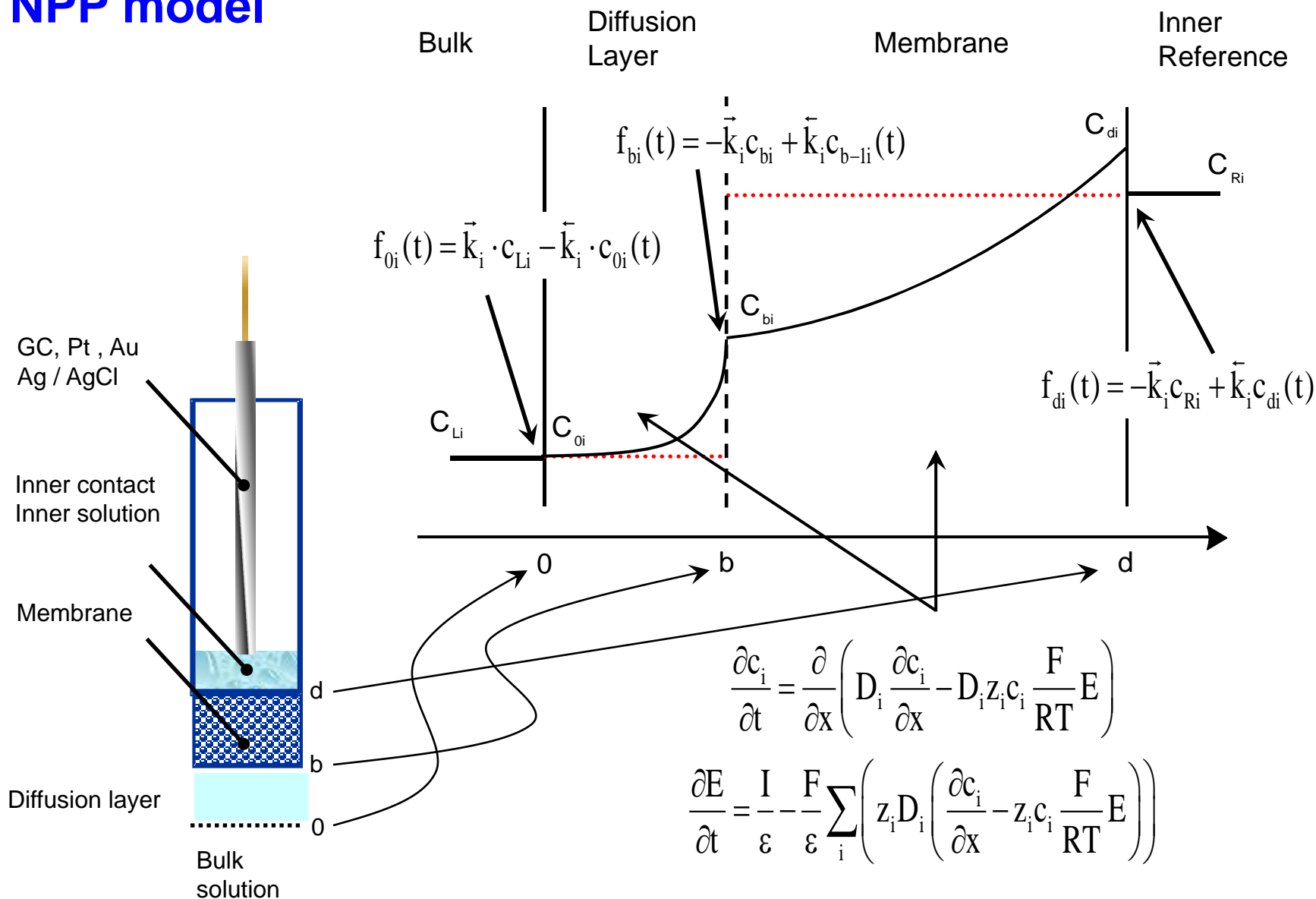


Walther Hermann **NERNST**  
1864-1941



Max **PLANCK**  
1858-1947

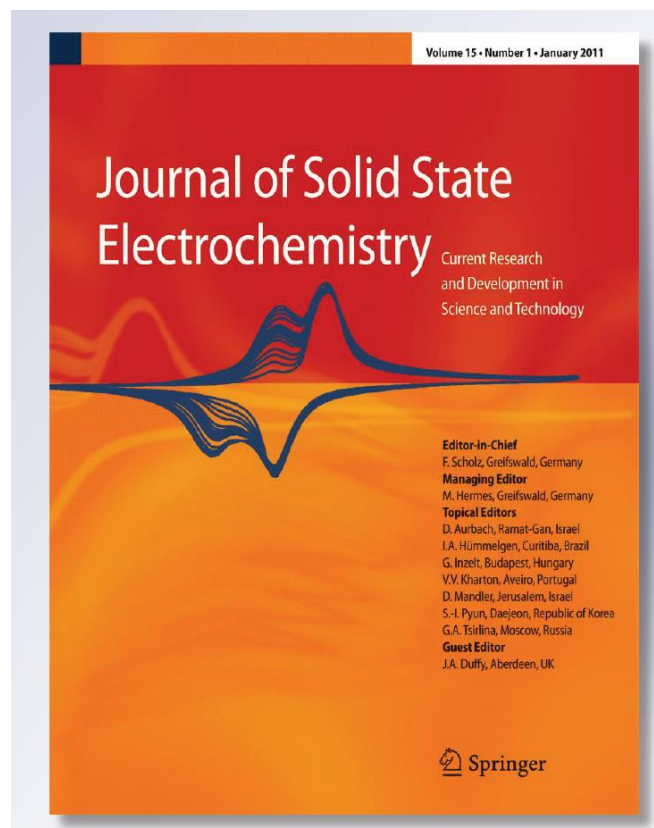
# NPP model



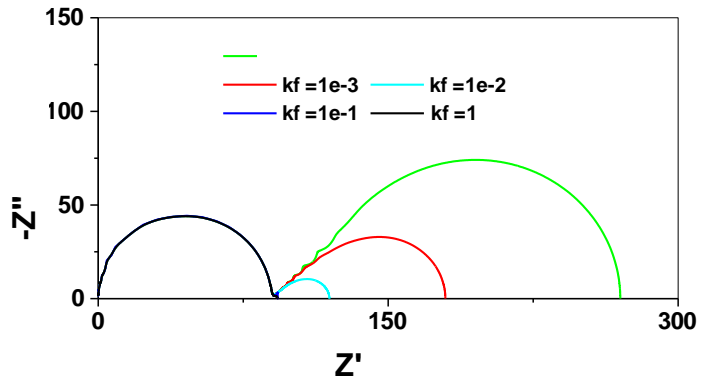
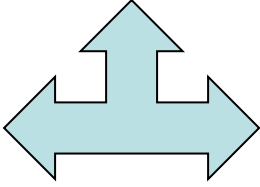
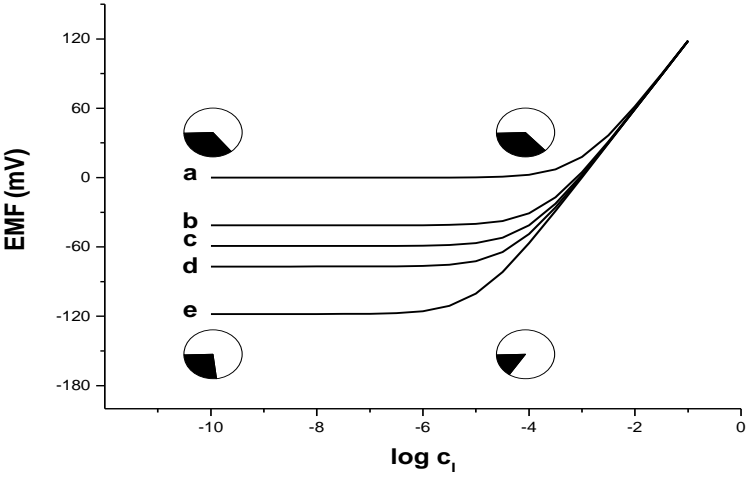
REVIEW

# Non-equilibrium potentiometry—the very essence

Andrzej Lewenstam



# Non equilibrium potentiometry: NPP 'general' power



$$E_m = f(a_i, a_j, a_p, d, \varepsilon, \delta, D_i, D_j, D_p, k_i, k_j, k_p, x, t)$$

$$I \text{ or } Z = f(\dots) \text{ AND INVERSE}$$





**JES FOCUS ISSUE ON MATHEMATICAL MODELING OF ELECTROCHEMICAL SYSTEMS AT MULTIPLE SCALES IN HONOR OF JOHN NEWMAN**

## **Modeling of Electrodifusion Processes from Nano to Macro Scale**

**K. Szyszkiewicz, J. J. Jasielc, M. Danielewski, A. Lewenstam, and R. Filipek<sup>z</sup>**

*Faculty of Materials Science and Ceramics, AGH University of Science and Technology, 30-059 Kraków, Poland*

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Mass and charge transport processes play an important role in different areas of science. The membrane processes involving charge transport are of vital importance in cell biology since they support homeostasis of living organisms. In the field of ion-selective electrodes transport of ions determines generation of electrochemical signals. The process of ionic diffusion remains of primary importance in many civil engineering problems since the long-term durability of building materials, such as concrete, is directly affected by the transport of chemical species. Length scale of the above processes range from nanometers for biological membranes through micrometers for ion selective electrodes up to centimeters for ions transport in concrete.

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All rights reserved.



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Manuscript submitted April 3, 2017; revised manuscript received June 23, 2017. Published July 13, 2017. *This paper is part of the JES Focus Issue on Mathematical Modeling of Electrochemical Systems at Multiple Scales in Honor of John Newman.*

## CHALLENGES...cont.

Membranes glass, crystal (solid-state), liquid (plastic), gas sensors, biosensors (metabolites, enzyme)

## New parameters...new membranes

...ionized  $Mg^{2+}$

(1992)

...bicarbonates  $HCO_3^-$

(2019)

### A historical success: the measurement of ionized magnesium

Andrzej Lewenstam, Head of Sensor Technology Group at Kone Instruments, announced the possibility of fully automatic determination of ionized magnesium during the 11th Meeting of International Federation of Clinical Chemistry on "Methodology and clinical application of ion-selective electrodes" held in Monterey, California, this summer.



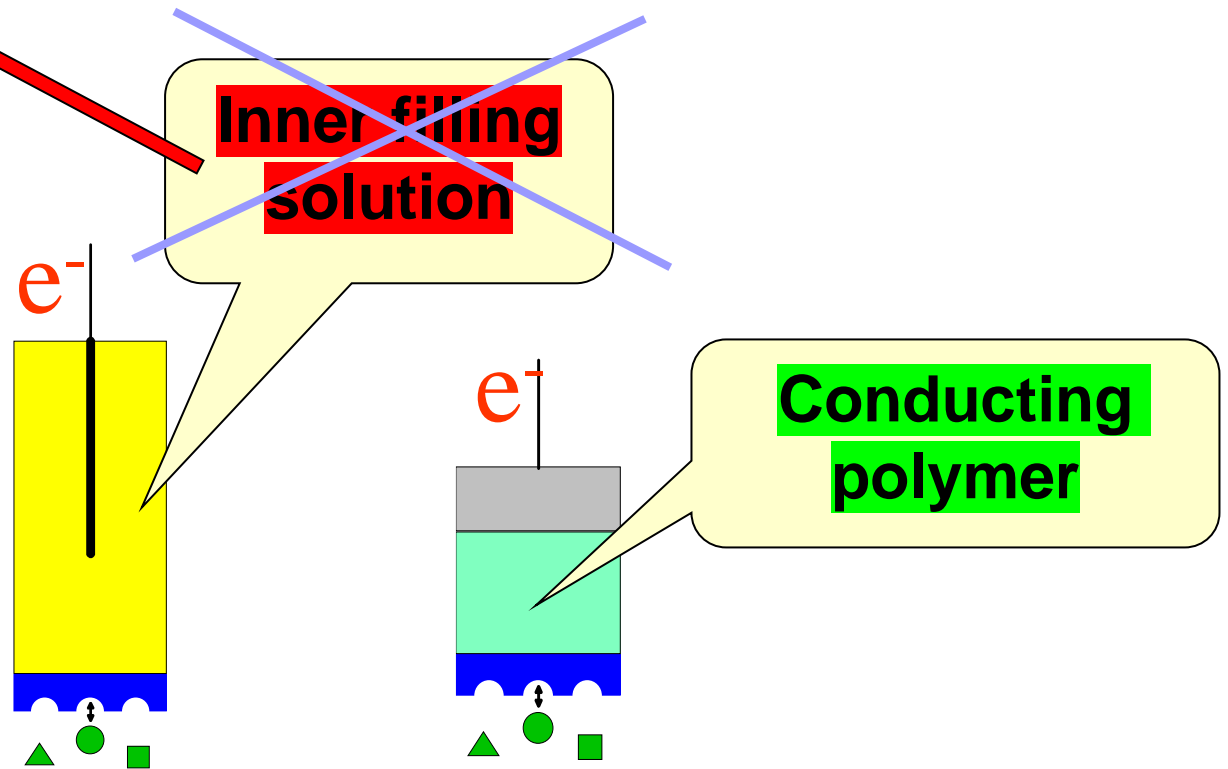
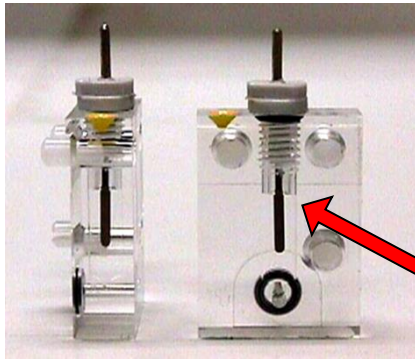
Andrzej Lewenstam for once not photographed in laboratory surroundings.

A Breakthrough Application of a Cross-Linked Polystyrene Anion-Exchange Membrane for a Hydrogencarbonate Ion-Selective Electrode

S. Dabrowska, J. Migdalski, A. Lewenstam

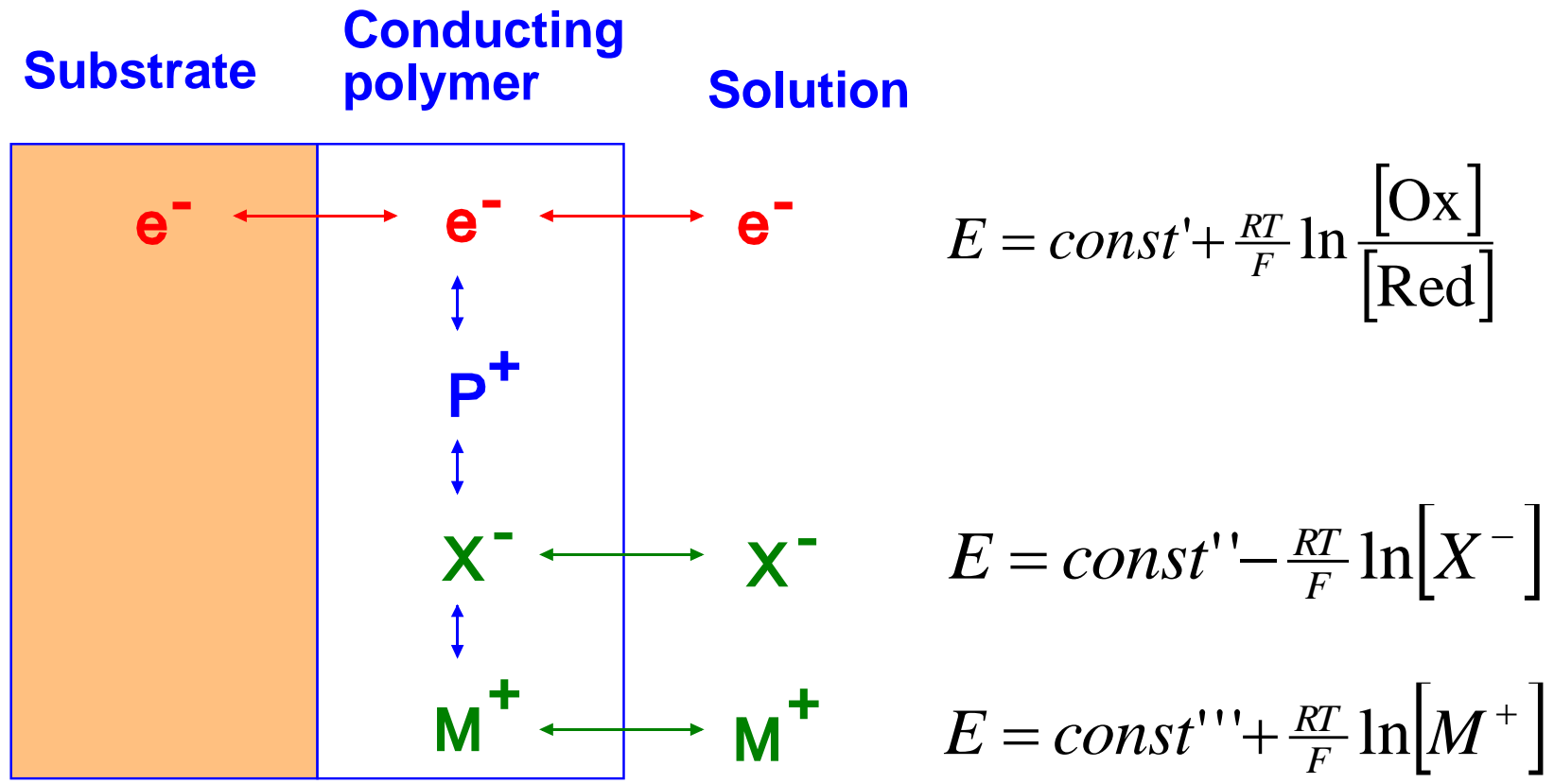
*Sensors*, **2019**, 19(6), 1268

# Solid-contact ion-sensors



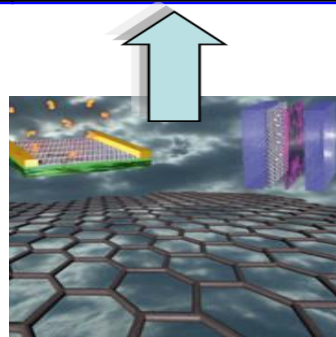
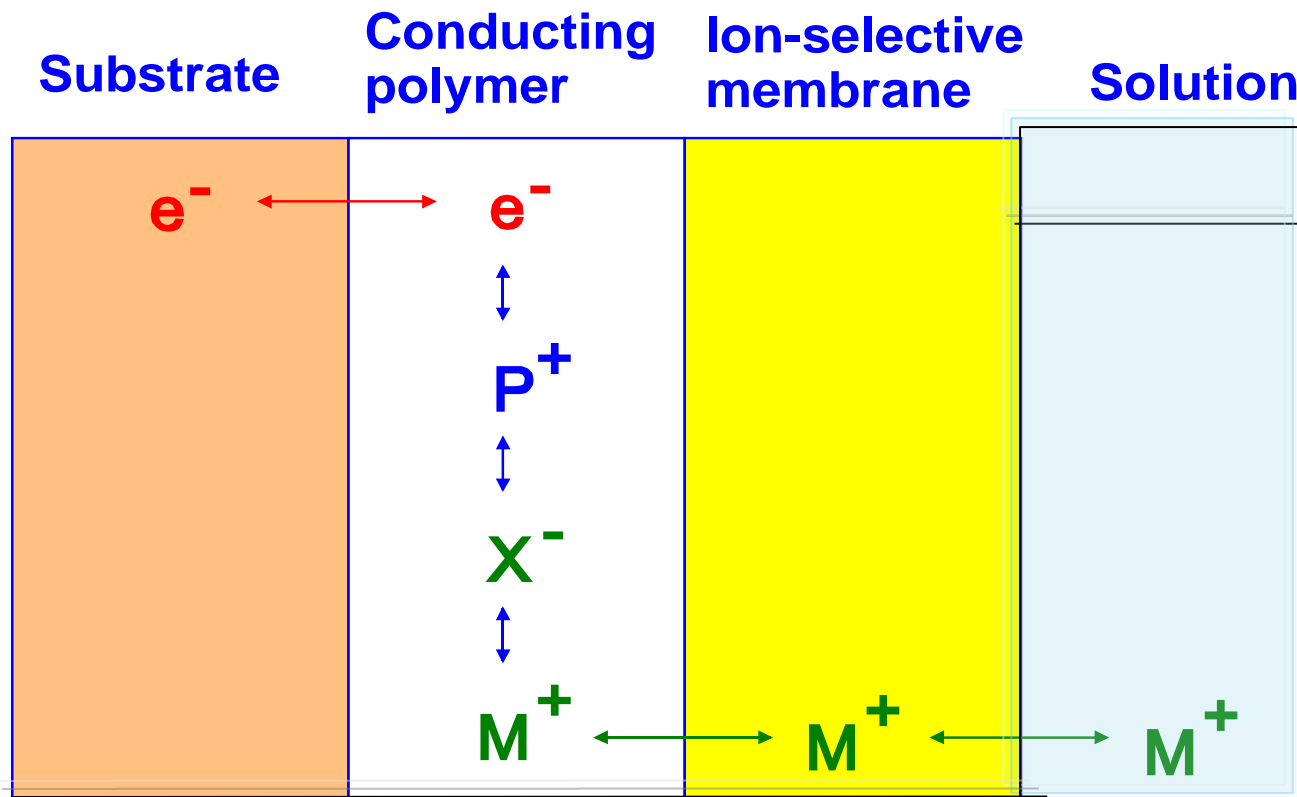
# Ion-to-electron transduction by conducting polymer

Generalized interpretation of a potentiometric (open-circuit) response of conducting polymer



A. Lewenstam et al. J. Electroanal. Chem., 368 (1994) 23-31.

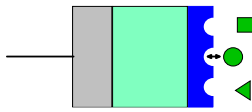
# *Solid contact all-solid-state ion-sensor*



**Graphene**

**Graphene:** Nobel 2010  
**Conducting polymer:** Nobel 2000

B. Paczosa-Bator et al, AGH  
 A. Michalska et al, UW



interface I

interface II

SUBSTRATE

SOLID CONTACT

MEMBRANE/BRIDGE

electron

ion(s)

coupling

electron-to-ion transducer

Solid contact as ion-to-electron transducer. Ion-to-electron coupling scheme.

...toward fully integrated galvanic probes...

György Inzelt · Andrzej Lewenstam · Fritz Scholz *Editors*

## Handbook of Reference Electrodes

Reference Electrodes are a crucial part of any electrochemical system, yet an up-to-date and comprehensive handbook is long overdue. Here, an experienced team of electrochemists provides an in-depth source of information and data for the proper choice and construction of reference electrodes. This includes all kinds of applications such as aqueous and non-aqueous solutions, ionic liquids, glass melts, solid electrolyte systems, and membrane electrodes. Advanced technologies such as miniaturized, conducting-polymer-based, screen-printed or disposable reference electrodes are also covered. Essential know-how is clearly presented and illustrated with almost 200 figures.

Inzelt · Lewenstam  
Scholz *Eds.*

György Inzelt  
Andrzej Lewenstam  
Fritz Scholz *Editors*



Handbook of Reference Electrodes

# Handbook of Reference Electrodes

ISBN 978-3-642-36187-6

Chemistry

ISBN 978-3-642-36187-6



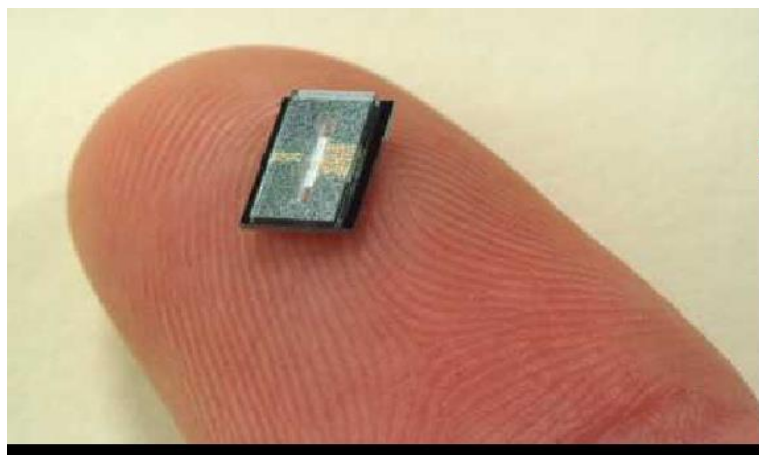
9 783642 361876

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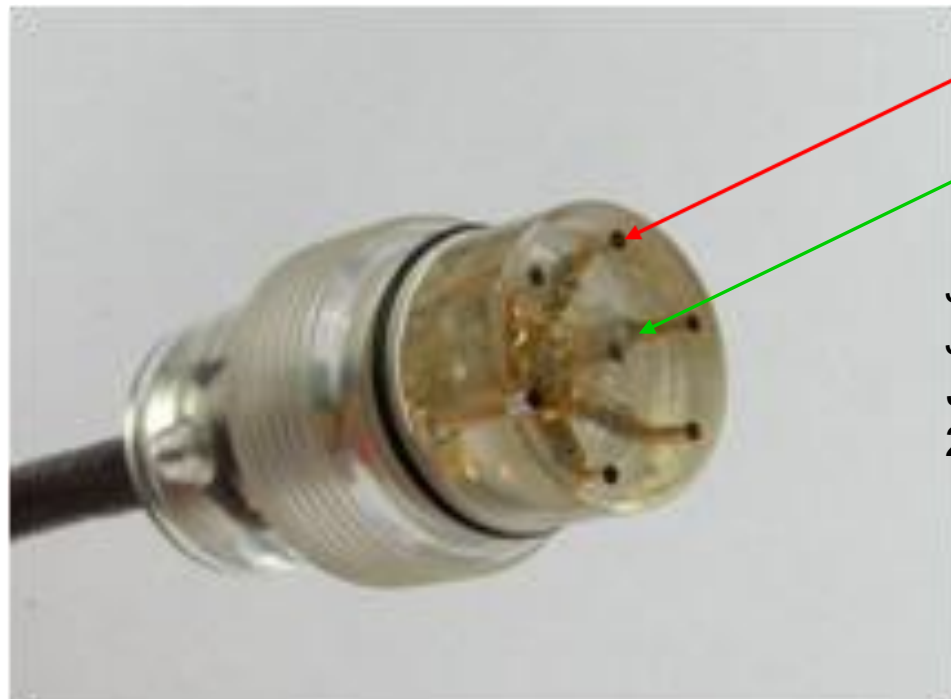
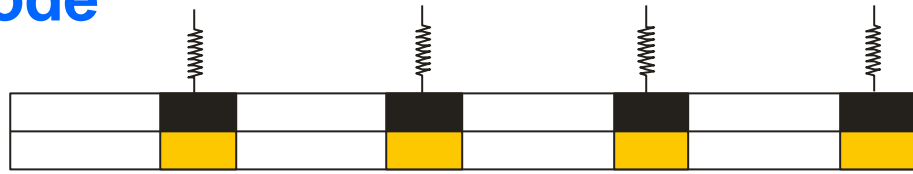
 Springer



**Solid contact ISE  
All-solid-state ISE  
& REF electrode**



**...multielectrode  
platform**



ISE

REF

J. Migdalski, B. Bas, T. Blaz,  
J. Golimowski, A. Lewenstam  
*J. Solid State Electrochem.*,  
2009, **13**, 149

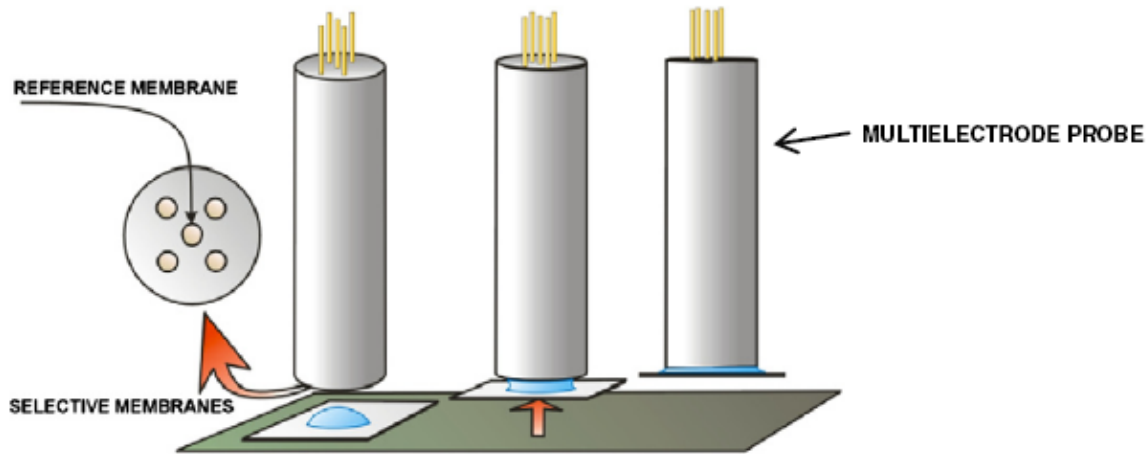


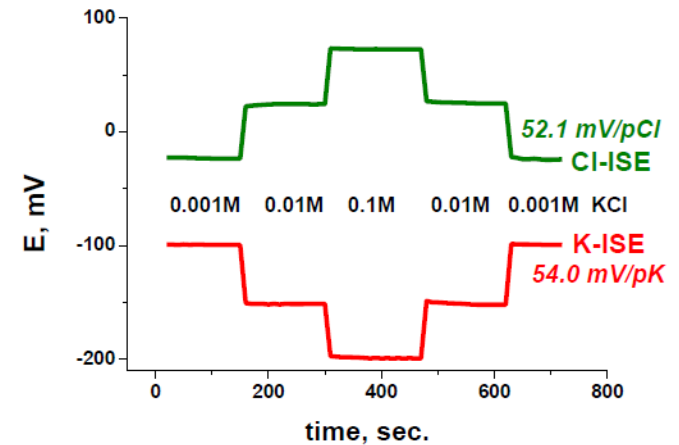
Fig. 1. Principle of "contact-type" one-drop measurement.



## Multielectrode potentiometry in a one-drop sample

T. Blaz, B. Baś, J. Kupis, J. Migdalski, A. Lewenstam \*

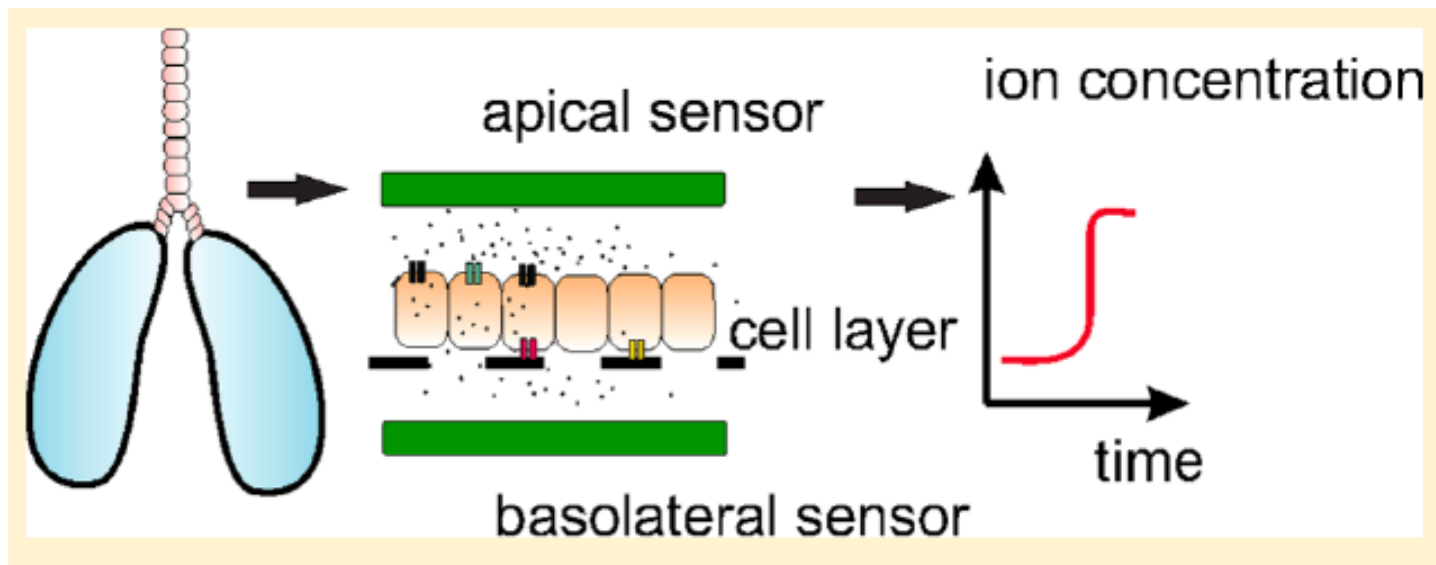
Electrochemistry Communications 34 (2013) 181–184



# Multielectrode Bisensor System for Time-Resolved Monitoring of Ion Transport Across an Epithelial Cell Layer

Renata Toczyłowska-Mamińska,<sup>†</sup> Andrzej Lewenstam,<sup>\*,‡</sup> and Krzysztof Dołowy<sup>†</sup>

*Anal. Chem.* 2014, 86, 390–394



....a human bronchial epithelium cell line 16HBE14 $\sigma$ - was used

# **Fabrication of Electrodes** **by 3D Printing**

# Solid contact reference electrode with a PVC-based composite electroactive element fabricated by 3D printing

A. Lewenstam, B. Bartoszewicz, J. Migdalski, A. Kochan

Electrochemistry Communications 109 (2019) 106613

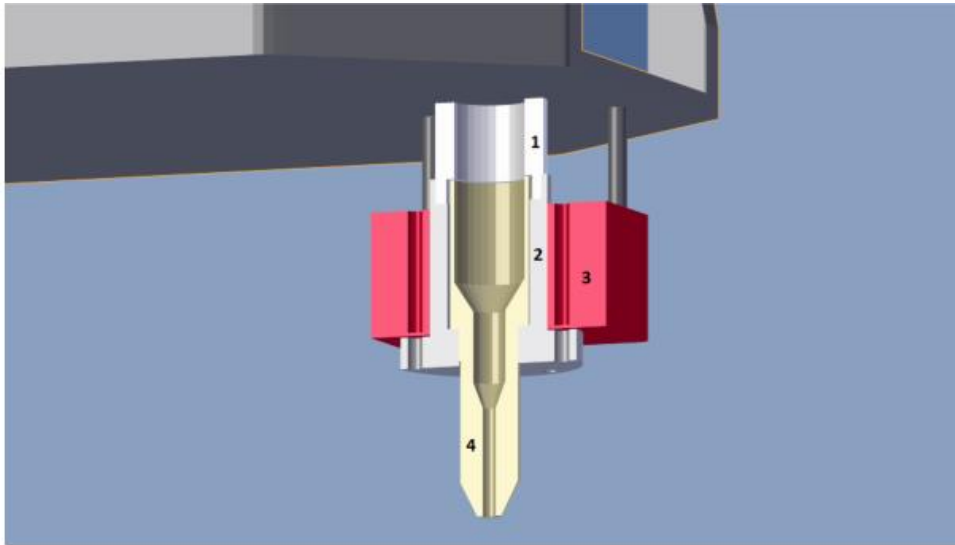


Fig. 1. Cross-section of the nozzle area of a composite printing module. 1 – PTFE heat break, 2 – aluminum heat link, 3 – aluminum heat block, 4 – brass nozzle additional intermediate compression zone.



Fig. 2. Solid state 3D printed reference electrode. On the left-side a scheme with electrode parts; on the right-side fabrication steps and photos of a real product (dimensions of the base: 30 mm × 20 mm and cavity diameter 11 mm).

# Applications...

- ...environmental analysis**
- ...process control**
- ...point-of-care measurements**
- ...wearable sensors**



## Multisensor Probe

### Clinical Analysis:



**Blood & Urine:**  
 $K^+$ ,  $Na^+$ ,  $Cl^-$ , pH etc.



### Environment:



**Surface Water & Soil:**  
 $Pb^{2+}$ ,  $Cd^{2+}$ , pH etc.

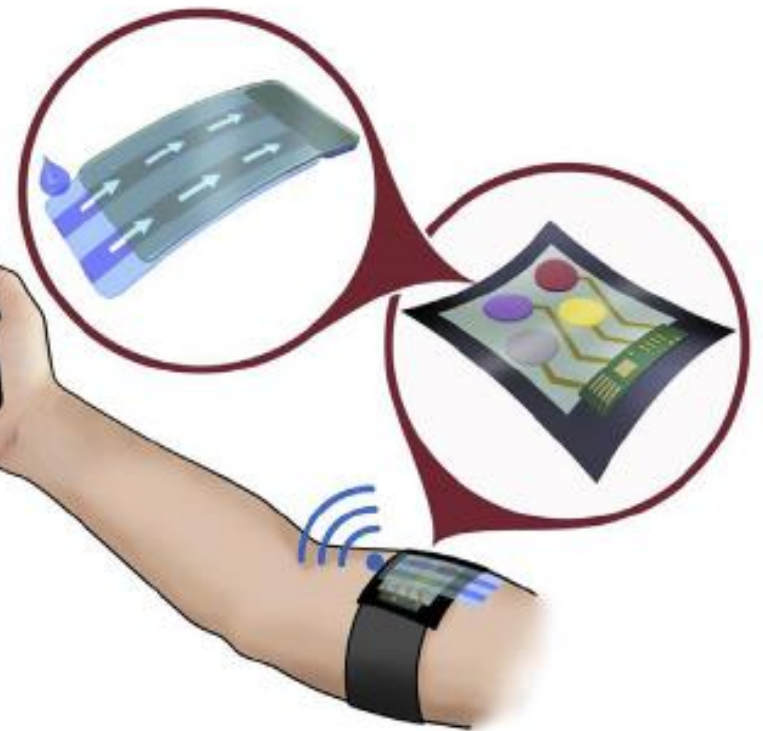
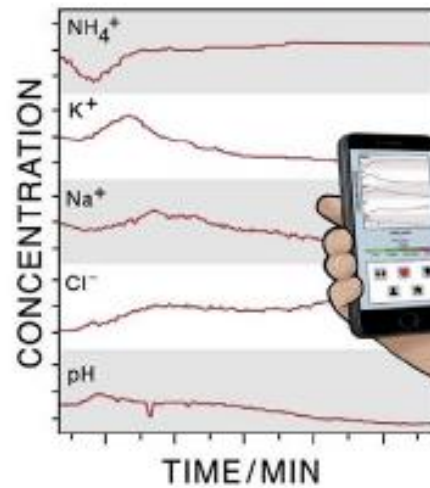


### Industry:

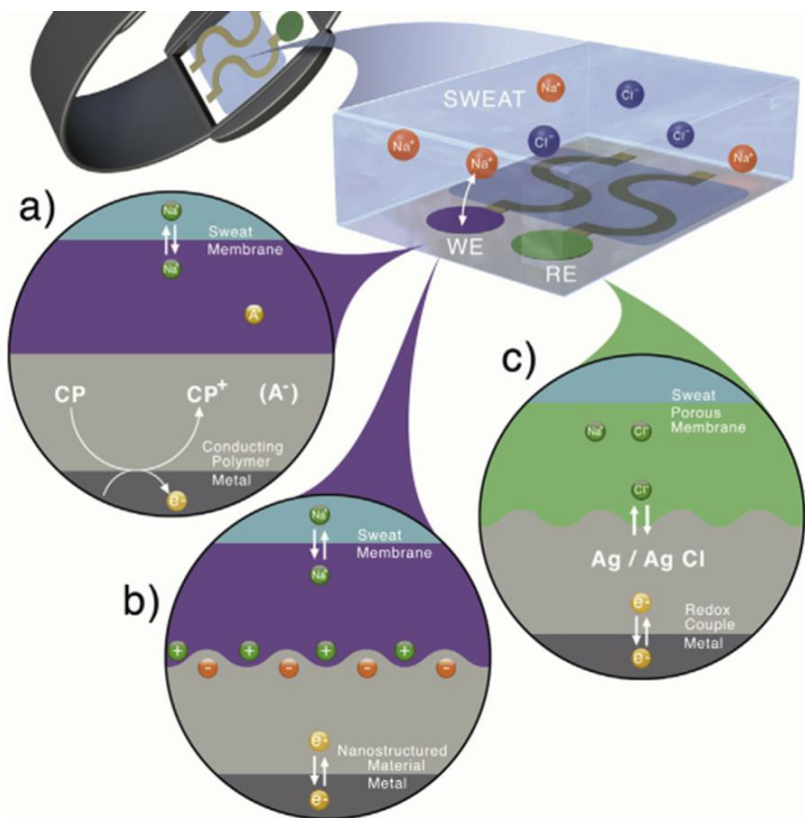


**Process Chemicals & Liquors:**  
 $Ca^{2+}$ ,  $Fe^{2+}$ , pH etc.



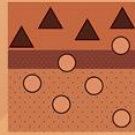


...potentiometric ion-sensors as health quality controllers.



...solid contact wearable ion-sensors





*membranes*

# Advances in Artificial and Biological Membranes

Mechanisms of Ionic Sensitivity,  
Ion-Sensor Designs and Applications  
for Ions Measurement

Edited by

Andrzej Lewenstam and Krzysztof Dołowy

Printed Edition of the Special Issue Published in *Membranes*

# SUMMARY

*From pH measurements to automated clinical analysis ...*

